

DESCRIPTION OF VIBRATIONAL QUANTUM NUMBER IMPLEMENTATIONS IN FIELDS E1 and E2 for GEISA-2020

V_j: quantum number associated with the normal mode of vibration j

$\sum \mathbf{xV}_j$: combination vibrational state

l_j: vibrational angular momentum quantum number associated with the degenerate bending mode j

l: absolute value of the sum of the vibrational angular momentum quantum number l_j .

GROUND for identification of fundamental vibrational state of some specific molecules, i.e.:

HNO₃, C₂H₆, CH₃D, C₂H₄, C₃H₈, CH₃Cl, HCOOH, SF₆, C₃H₄, ClONO₂, CH₃Br, CH₃OH, C₆H₆, CH₃CN, ...

GREY LINE if no quantum identification of vibrational levels

Molecule in GEISA-2015:

H₂O, CO₂, O₃, N₂O, CO, CH₄, O₂, NO, SO₂, NO₂, NH₃, PH₃, HNO₃, OH, HF, HCl, HBr, HI, ClO, OCS, H₂CO, C₂H₆, CH₃D, C₂H₂, C₂H₄, GeH₄, HCN, C₃H₈, C₂N₂, C₄H₂, HC₃N, HOCl, N₂, CH₃Cl, H₂O₂, H₂S, HCOOH, COF₂, SF₆, C₃H₄, HO₂, ClONO₂, CH₃Br, CH₃OH, NO⁺, HNC, C₆H₆, C₂HD, CF₄, CH₃CN, HDO, SO₃

New molecule in GEISA-2020:

HONO, COFCl, CH₃I, CH₃F, RUO₄, H₂C₃H₂

References:

- [1] Tashkun et al., JQSRT 152 (2015) 45–73
- [2] Barbe et al., JQSRT 130 (2013) 172–190
- [3] Brown et al., JQSRT 130 (2013) 201–219
- [4] Down et al., JQSRT 130 (2013) 260–272
- [5] Gomez et al., JQSRT 111 (2010) 2256–2264

[6] Perrin et al. JMS 171 (1995) 358-373

H₂O Molecule 1

V1,V2,V3

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
																				◆		◆		◆

E2: lower state vibrational identification

																				◆		◆		◆
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CO₂ Molecule 2

V1,V2,l2,V3,r

r: ranking index[1]

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
																	◆		◆		◆		◆	◆

E2: lower state vibrational identification

																	◆		◆		◆		◆	◆
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O₃ Molecule 3

V1,V2,V3_r

r: ranking index [2]

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																					❖	❖	❖	-	❖

E2: lower state vibrational identification

																					❖	❖	❖	-	❖
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N₂O Molecule 4

V1,V2, l2,V3

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																						❖	❖	❖	❖

E2: lower state vibrational identification

																						❖	❖	❖	❖
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O₂ Molecule 7

X,V1

X electronic state of the molecule

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																								❖	❖

E2: lower state vibrational identification

																									❖	❖
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NO Molecule 8

X,i,V1

X electronic state of the molecule; i=1/2 or 3/2

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																	❖	❖	❖	❖				❖	❖

E2: lower state vibrational identification

																	❖	❖	❖	❖				❖	❖
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SO₂ Molecule 9

V1,V2,V3

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																					◆		◆		◆

E2: lower state vibrational identification

																					◆		◆		◆
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NO₂ Molecule 10

V1,V2,V3

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																							◆	◆	◆

E2: lower state vibrational identification

																							◆	◆	◆
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HNO₃ Molecule 13

$$\sum xV_j$$

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																					❖	❖	❖	❖	❖
																				G	R	O	U	N	D

E2: lower state vibrational identification

																							❖	❖	❖
																				G	R	O	U	N	D

OH Molecule 14

$$X, i, V1$$

X electronic state of the molecule; i=1/2 or 3/2

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																									❖

E2: lower state vibrational identification

CIO Molecule 19

X,i,V1

X electronic state of the molecule; $i=1/2$ or $3/2$

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
																	❖	❖	❖	❖			❖	❖

E2: lower state vibrational identification

																	❖	❖	❖	❖			❖	❖
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OCS Molecule 20

V1,V2,I2,V3

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																						❖	❖	❖	❖

E2: lower state vibrational identification

																						❖	❖	❖	❖
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H₂CO Molecule 21

V1,V2,V3,V4,V5,V6

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
																			❖	❖	❖	❖	❖	❖

E2: lower state vibrational identification

																			❖	❖	❖	❖	❖	❖
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C₂H₆ Molecule 22

$\sum xV_j$

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
															❖	❖	❖	❖	❖	❖	❖	❖	❖	❖
																			G	R	O	U	N	D

E2: lower state vibrational identification

																						❖	❖	❖
																			G	R	O	U	N	D

CH₃D Molecule 23

$$\sum xV_j$$

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
																	❖	❖	❖	❖	❖	❖	❖	❖
																			G	R	O	U	N	D

E2: lower state vibrational identification

																				G	R	O	U	N	D

C₂H₂ Molecule 24

V₁,V₂,V₃,V₄,V₅,l,+/-,r,p +/-: Symmetry type for \sum levels (l=0) r: ranking index; p: vibrational level u or g parity [5]

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
																❖	❖	❖	❖	❖	❖	❖	❖	❖

E2: lower state vibrational identification

																	❖	❖	❖	❖	❖	❖	❖		❖
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C₂H₄ Molecule 25

$$\sum xV_j$$

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																		❖	❖	❖	❖	❖	❖	❖	❖

E2: lower state vibrational identification

																								❖	❖	❖					
																										G	R	O	U	N	D

GeH₄ Molecule 26

$$V_1, V_2, I_2, V_3, I_3, V_4, I_4, I$$

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																		❖	❖	❖	❖	❖	❖	❖	❖

HC₃N Molecule 31

V1,V2,V3,V4,V5,V6,V7,I5,I6,I7,S

S: Vibrational symmetry

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
					❖		❖		❖		❖		❖		❖		❖	❖	❖	❖	❖	❖	❖	❖

E2: lower state vibrational identification

					❖		❖		❖		❖		❖		❖		❖	❖	❖	❖	❖	❖	❖	❖
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HOCl Molecule 32

V1,V2,V3

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																							❖	❖	❖

E2: lower state vibrational identification

																							❖	❖	❖
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N₂ Molecule 33

V_j

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																									❖

E2: lower state vibrational identification

																										❖
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CH₃Cl Molecule 34

∑ xV_j

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																				❖	❖	❖	❖	❖	❖
																				G	R	O	U	N	D

E2: lower state vibrational identification

																				G	R	O	U	N	D
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H₂O₂ Molecule 35

V1,V2,V3,n4,t4,V5,V6

n4,t4: torsional quantum numbers [6]

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
														❖		❖		❖	❖	❖		❖		❖

E2: lower state vibrational identification

														❖		❖		❖	❖	❖		❖		❖
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H₂S Molecule 36

V1,V2,V3

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																							❖	❖	❖

E2: lower state vibrational identification

																							❖	❖	❖
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HCOOH Molecule 37

$$\sum xV_j$$

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																				❖	❖	❖	❖	❖	
																				G	R	O	U	N	D

E2: lower state vibrational identification

																				G	R	O	U	N	D
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COF₂ Molecule 38

V1,V2,V3,V4,V5,V6

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																				❖	❖	❖	❖	❖	❖

E2: lower state vibrational identification

																				❖	❖	❖	❖	❖	❖
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SF₆ Molecule 39 $\sum xV_j, S, p$ *S*: vibrational symmetry; *p*: parity g or u**E1: upper state vibrational identification**

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
																❖	❖	❖	❖	❖		❖	❖	❖

E2: lower state vibrational identification

																			❖	❖		❖	❖	❖
																			G	R	O	U	N	D

C₃H₄ Molecule 40

V_j or V_j HOT L=x or As₁As₂V_j HOT L=x

s1 or s2: +/-

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																							❖	❖	❖
										❖	❖	❖	❖	❖	❖	❖		❖	❖	❖		❖	❖	❖	

E2: lower state vibrational identification

																			G	R	O	U	N	D

HO₂ Molecule 41

V₁,V₂,V₃

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																							❖	❖	❖

E2: lower state vibrational identification

																									❖	❖	❖
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ClONO₂ Molecule 42

$$\sum xV_j$$

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																					❖	❖	❖	❖	❖
																				G	R	O	U	N	D

E2: lower state vibrational identification

																									❖	❖
																				G	R	O	U	N	D	

CH₃Br Molecule 43

$$V_j$$

NO+ Molecule 45

Vj

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																									❖

E2: lower state vibrational identification

																										❖
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HNC Molecule 46

V1,V2,l2,V3

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
																		❖	❖	❖		❖		❖

E2: lower state vibrational identification

																		❖	❖	❖		❖		❖
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C₆H₆ Molecule 47

Vj

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																								❖	❖

E2: lower state vibrational identification

																									G	R	O	U	N	D
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C₂HD Molecule 48

V1,V2,V3,V4,V5,I4,I5,S

S: vibrational symmetry

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25				
										❖		❖		❖		❖		❖	❖	❖	❖	❖	❖	❖	❖	❖	❖	❖

E2: lower state vibrational identification

										❖		❖		❖		❖		❖	❖	❖	❖	❖	❖	❖	❖	❖	❖	❖
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CF₄ Molecule 49

V1,V2,V3,V4,n,C

n: multiplicity index; C: vibrational symmetry

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25
																	❖	❖	❖	❖		❖	❖	❖

E2: lower state vibrational identification

																		❖	❖	❖	❖		❖	❖	❖
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CH₃CN Molecule 50

V_j

E1: upper state vibrational identification

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	
																								❖	❖

E2: lower state vibrational identification

																				G	R	O	U	N	D

HDO Molecule 51

V₁,V₂,V₃

